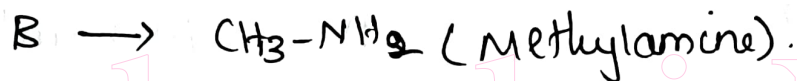
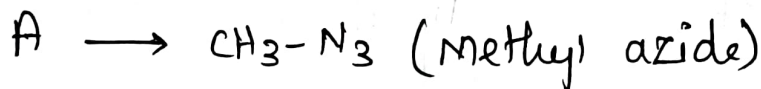
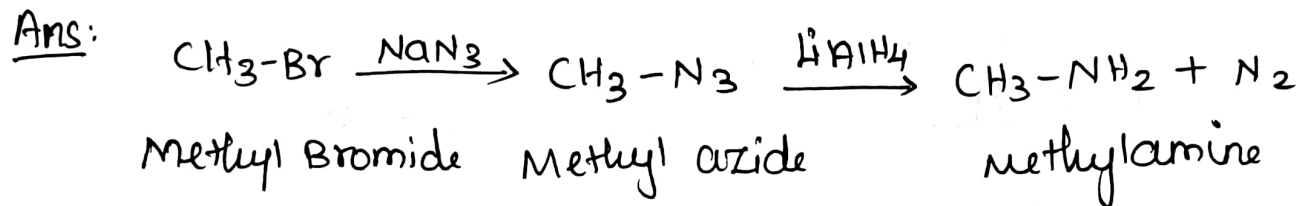


CHEMISTRY12th - Public Examination 2022

ONE MARK - ANSWER KEY :

TYPE - A		TYPE - B	
1)	C - $1s^2 2s^2 2p^6 3s^2 3p^3$	1)	C - Glycine
2)	A - $Al_2O_3 \cdot NH_2O$	2)	A - Methanal
3)	D - 30 minutes	3)	A - Basic
4)	C - Sn/HCl	4)	D - FeO
5)	A - Methanal	5)	C - Hydrolysis of sucrose in presence of dil. HCl.
6)	C - $[Fe(CO)_5]$	6)	C - Sn/HCl
7)	A - HPO_4^{2-}	7)	D - charge carried by one mole of electrons
8)	A - Basic	8)	A - $Al_2O_3 \cdot NH_2O$
9)	C - Hydrolysis of sucrose in presence of dil. HCl.	9)	D - 30 minutes
10)	C - Nucleophilic Addition	10)	A - HPO_4^{2-}
11)	C - +3	11)	C - $[Fe(CO)_5]$
12)	D - charge carried by one mole of electrons.	12)	C - +3
13)	C - Glycine	13)	C - $1s^2 2s^2 2p^6 3s^2 3p^3$
14)	D - FeO	14)	A - Liquid in gas
15)	A - Liquid in gas	15)	C - Nucleophilic Addition

2 MARK - Compulsory:



3 MARK Compulsory:

