LESSON 1 HEAT

Answer briefly.

1. What are the applications of conduction in our daily life?

- We cook food in vessels made up of metals. When the vessel is heated, heat is transferred from the metal to the food.
- When we iron dresses heat is transferred from the iron to the cloth.

2. What are the effects of heat?

- ➢ Expansion
- Increase in temperature
- ➢ Change in state

3. Name three types of heat transfer.

- Conduction
- Convection
- ➢ Radiation

4. What is conduction?

- In conduction heat transfer takes place between two ends of the same solid or through two solid substances that are at different temperatures but in contact with one another.
- 5. Write a note on convection.
 - When water in the vessel is heated, water molecules at the bottom receive heat energy and move upward.
 - Then the molecules at the top comes down and get heated. This kind of heat transfer is known as convection.

6. Define specific heat capacity.

- Specific heat capacity of a substance is defined as the amount of heat energy required to raise the temperature of 1 kilogram of a substance by 1°C or 1 K.
- ➢ It is denoted by the symbol C.

OR

Specific heat capacity = $\frac{\text{Amount of heat energy required (Q)}}{\text{Mass} \times \text{Raise in temperature (}\Delta\text{T)}}$

7. Define one calorie.

- Calorie is defined as the amount of heat energy required by a substance to raise its temperature by 1°C or 1 K.
 - It is denoted by the symbol C'

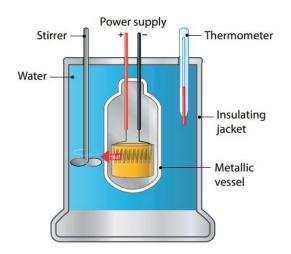
OR

Calorie = $\frac{\text{Amount of heat energy required (Q)}}{\text{Raise in temperature }(\Delta T)}$

VII. Answer in detail.

1. With the help of a neat diagram explain the working of a calorimeter.

A calorimeter is a device used to measure the amount of heat gained or lost by a substance.



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WORKING:

- The metallic vessel is kept in an insulating jacket to prevent heat loss.
- Through one hole a thermometer is inserted to measure the Heat temperature of the contents.
- A stirrer is inserted through another hole for stirring the content in the vessel.
- The vessel is filled with liquid which is heated by passing current through the heating element.
- Using this device we can measure the heat capacity of the liquid in the container.

2. Write a note on thermostat.

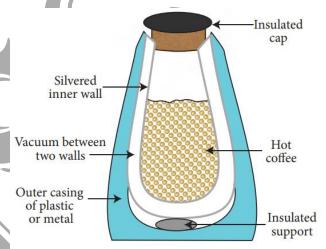
- A thermostat is a device which maintains the temperature of a place or an object constant.
- Greek word thermo means heat and static means staying the same.
- Thermostats are used in any device or system that gets heated or cools down to a pre-set temperature.
- Sometimes, a thermostat functions both as the sensor and the controller of a thermal system.
- E.g: Air conditioner, water heater.

3. Explain the working of thermos flask.

- The thermos flask (Vacuum flask) is an insulating storage vessel
- it keeps its content hotter or cooler than the surroundings for a longer time.

Working of Thermos flask:

- A thermos flask has double walls, It is silvered on the inside.
- The vacuum between the two walls prevents heat being transferred from the inside to the outside by conduction and convection.
- The silvered walls reflect radiated heat back to the liquid in the bottle.



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LESSON 02 - ELECTRICITY

Answer briefly.

1. How charges are produced by friction?

- Comb rubbed with hair becomes negatively charged.
- Positive and negative charges are present at the edges of the paper.
- Negative charges in the comb attract positive charges in the bits

2. What is earthing?

A safety measure devised to prevent people from getting shocked if the insulation inside electrical devices fails is called Earthing.

3. What is electric circuit?

The path through which electrons flow from one terminal to another terminal of the source, is called electric circuit.

4. What is electroplating?

The process of depositing a layer of one metal over the surface of another metal by passing electric current in called electroplating.

5. Give some uses of electroplating.

- ➢ Car parts,
- ➢ Bath taps,
- Bicycle handlebars,
- Wheel rims

VIII. Answer in detail.

1. Explain three ways of charge transfer.

Transfer of charges takes place in the following three ways.

Transfer by Friction

In this type transfer of charges by friction.
 E.g: paper bits and comb

Transfer by Conduction

➢ In this type transfer of charges by

conduction. E.g: ebonite rod and woolen

cloth.

Transfer by Induction

- In this type transfer of charges by induction. E.g: we can charge an uncharged object
- 2. What is electroscope? Explain how it works?
 An electroscope is a scientific instrument used to detect the presence of electric
 - charge on a body
 - In the year 1600, British physician William
 Gilbert invented the first electroscope.
 - There are two types of electroscope: pithball electroscope and goldleaf electroscope.

Working of Electroscope

- When the brass disc of the electroscope is touched by a charged object, electric charge gets transferred to the gold leaf through the rod.
- This results in the gold leaves moving away from each other.

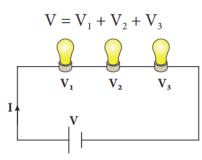
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This happens because both the leaves have similar charges.

3. Explain series and parallel circuit.

Series Circuit

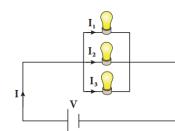
A series circuit is one that has more than one resistor (bulb) but only one path through which the electrons can travel.



Parallel Circuit

In a parallel circuit, there is more than one resistor (bulb) and they are arranged on many paths.

 $\mathbf{I} = \mathbf{I}_1 + \mathbf{I}_2 + \mathbf{I}_3$



4. How lightning takes place?

- Lightning is produced by discharge of electricity from cloud to cloud or from cloud to ground.
- During thunderstorm air is moving upward rapidly. This air which moves rapidly, carries small ice crystals upward.
- The upper part of the cloud is positively charged and the lower part of the cloud is negatively charged.
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- When they come into contact, electricity is generated and lightning is seen.
- Sometimes lightning may be seen before the thunder is heard. This is because the distance between the clouds and the surface.

5. What is electroplating? Explain how it is done.

- The process of depositing a layer of one metal over the surface of another metal by passing electric current in called electroplating.
- When electric current is passed through the copper sulphate solution, you will find that a thin layer of copper metal is deposited on the iron spoon and an equivalent amount of copper is lost by the copper plate.

LESSON 03 AIR

- I. Answer briefly.
- 1. What are the sources of oxygen?
 - \blacktriangleright Atmospheric air
 - ➤ Water
- 2. Mention the physical properties of oxygen.
 - Oxygen is a colourless, odourless and tasteless gas.
 - \succ It supports combustion.
 - It is a poor conductor of heat and electricity

3. List out the uses of nitrogen.

- Liquid nitrogen is used as a refrigerant
- It is used for the preservation of fresh foods
- It is used for inflating tyres of vehicles.

4. Write about the reaction of nitrogen with non metals.

 N_2

Non-metal + Nitrogen \longrightarrow Nitrogen compound

3H, Nitrogen Hydrogen

2NH₃

Ammonia

5. What is global warming?

- ➤ The green house gases are CO2, N2O, CH4, CFC increases the temperature of earth's surface.
- \succ This is called global warming.
- 6. What is dry ice? What are its uses?
 - Solid carbon dioxide, called as dry ice is used as a refrigerant.

- ➢ CO2 can be used in the preservation of food grains, fruits etc.
- II. Answer in detail.

1. What happens when carbon dioxide is passed through lime water? Write the equation for this reaction.

Reaction with Lime water (Calcium) hydroxide) When a limited amount of CO2 is passed through lime water, it turns milky due to the formation of insoluble calcium

carbonate.

Ca(OH)2 + CO2

 \rightarrow CaCO3 + H2O Calcium carbonate

2. Name the compounds produced when the following substances burn in oxygen.

- a) Carbon b) Sulphur c) Phosphorous
- d) Magnesium e) Iron f) Sodium

Metal	Product
Carbon	Carbon di oxide
Sulphur	Sulphur di oxide
Phosphorous	Phosphorus trioxide
Magnesium	Magnesium oxide
Iron	Iron oxide
Sodium	Sodium oxide

3. How does carbon dioxide react with the

following? a) Magnesium b) Lime water

- c) Sodium hydroxide
- a) Magnesium

2Mg + Magnesium

 $CO_2 \longrightarrow 2MgO$ Magnesium oxide

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b) Lime water

 $Ca(OH)_2 + CO_2 \longrightarrow CaCO_3 + H_2O$ Calcium carbonate

c) Sodium hydroxide

 $2NaOH + CO_2 \longrightarrow Na_2CO_3 + H_2O$ Sodium carbonate

4. What are the effects of acid rain? How can we prevent them?

Effects of acid rain:

- > It irritates eyes and skin of human beings.
- > It changes the fertility of the soil.
- It inhibits germination and growth of seedlings.

Prevention:

- Minimizing the usage of fossil fuel such as petrol, diesel etc.,
- Using CNG (Compressed Natural Gas).
- ➢ Using non-conventional source of energy.
- > Proper disposal of the industrial wastes.

LESSON 04 ATOMIC

STRUCTURE

Answer briefly.

1. State the law of conservation of mass.

The law of conservation of mass states that during any chemical change, the total mass of the products is equal to the total mass of the reactants.

2. State the law of constant proportions.

 It states that in a pure chemical compound the elements are always present in definite proportions by mass.

3. Write the properties of anode rays.

- Anode rays travel in straight lines.
- Anode rays are made up of material particles.

4. Define valency of an element with respect to hydrogen.

- Valencies of the other elements are expressed in terms of hydrogen.
- valency of an element can also be defined as the number of hydrogen atoms which combine with one atom of it
- 5. Define the term ions or radicals.
 - Atoms which carry positive or negative charges are called ions.
 - An atom or a group of atoms when they either lose or gain electrons, get converted into ions or radicals

6. What is a chemical equation?

- > A chemical equation is a short hand
 - representation of a chemical reaction with

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formulae. formulae. a) C + O ₂ → CO ₂ b) P + Cl ₅ → PCl ₅ c) S + O ₂ → SO ₂ d) Mg + HCl ₂ → MgCl ₂ + H b) N ₂ O - Nitrogen oxide c) NO ₂ - Nitrogen di oxide d) PCl ₅ - Phosphorus penta chloride V. Answer the following. 1. Find the valency of the element which is underlined in the following formula. a) Na + O2 → Na ₂ O b) Ca + N ₂ → Ca ₃ N ₂ c) N ₂ + H ₂ → NH3 d) CaCO ₃ + HCl → Ca ₃ N ₂ c) N ₂ + H ₂ → NH3 d) CaCO ₃ + HCl → Ca ₂ Cl ₂ + O ₂ + H e) Pb(NO ₃) ₂ → Ca ₃ Cl ₂ + O ₂ + H e) Pb(NO ₃) ₂ → PbO + NO ₂ + O ₂ Answer: a) Na Cl b) CO ₂ c) <u>Al</u> (PO ₄) d) <u>Ba</u> (NO ₃) ₂ e) <u>Ca</u> Cl ₂ Answer: a) Na Cl b) <u>CO₂ c) Al</u> (PO ₄) d) <u>Ba</u> (NO ₃) ₂ e) <u>Ca</u> Cl ₂ Answer: a) Na Cl b) <u>CO₂ c) Al</u> (PO ₄) d) <u>Ba</u> (NO ₃) ₂ e) <u>Ca</u> Cl ₂ Answer: a) Na Cl b) <u>CO₂ c) Al</u> (PO ₄) d) <u>Ba</u> (NO ₃) ₂ e) <u>Ca</u> Cl ₂ Answer: a) Na Cl b) <u>CO₂ c) Al</u> (PO ₄) d) <u>Ba</u> (NO ₃) ₂ e) <u>Ca</u> Cl ₂ Answer: a) Alu(PO ₄) - 1 b) <u>Si</u> O ₂ - 2 2. Write the chemical formula for the following compounds a) Aluminium sulphate b) Silver nitrate c) Magnesium oxide d) Barium chloride Answer: a) Aluminium sulphate - Al ₃ SO ₄ b) Silver nitrate AgeNO ₃ c) Magnesium oxide - MgO ₂ d) Barium chloride BaCl ₂ 3. Write the skeleton equation for the following word equation and then balance them a) Carbon + Oxygen + Carbon dioside b) Phosphorus + Chlorine + Phosphorus pertachloride. b) Carbon + Oxygen + Carbon dioside b) Phosphorus + Chlorine + Phosphorus pertachloride. b) Carbon + Oxygen + Carbon dioside b) Phosphorus + Chlorine + Phosphorus pertachloride. b) Carbon + Oxygen + Carbon dioside b) Phosphorus + Chlorine + Phosphorus pertachloride. b) Carbon + Oxygen + Carbon dioside b) Phosphorus + Chlorine + Phosphorus pertachloride. b) Carbon + Coxygen + Carbon dioside b) Phosphorus + Chlorine + Phospho				R:	ANSWEF	ools and	mical symbol	elp of chem	the h	
a) CO b) N2O c) NO2 d) PCI5 a) CO - Carbon monoxide b) N ₂ O - Nitrogen oxide c) NO ₂ - Nitrogen di oxide d) PCl ₅ - Phosphorus penta chloride V. Answer the following. 1. Find the valency of the element which is underlined in the following formula. a) Na + O2 \rightarrow Na ₂ O b) Ca + N ₅ \rightarrow Ca ₃ N ₂ c) N ₂ + H ₂ \rightarrow NH3 d) CaCO ₃ + HCl \rightarrow CaCl ₂ + CO ₂ + H e) Pb(NO ₃) ₂ \rightarrow PbO + NO ₂ + O ₂ Answer: a) Na ^C L $-$ 1 b) CO ₂ c) Al (PO ₄) d) Ba (NO ₃) ₂ e) CaCl ₂ Answer: a) Na ^C L $-$ 1 b) CO ₂ $-$ 2 c) N ₂ + H ₂ \rightarrow NH3 d) CaCO ₃ + HCl \rightarrow CaCl ₂ + CO ₂ + H e) Pb(NO ₃) ₂ \rightarrow PbO + NO ₂ + O ₂ ANSWER: a) Na + O ₂ \rightarrow Na ₄ O b) 3Ca + N ₂ \rightarrow Ca ₃ N ₂ c) N ₂ + H ₂ \rightarrow NH3 d) CaCO ₃ + HCl \rightarrow CaCl ₂ + CO ₂ + H e) Pb(NO ₃) ₂ \rightarrow PbO + NO ₂ + O ₂ ANSWER: a) Na + O ₂ \rightarrow Na ₄ O b) 3Ca + N ₂ \rightarrow CaCl ₂ + CO ₂ + H e) Pb(NO ₃) ₂ \rightarrow PbO + 2NO ₂ + O ₂ for CaCl ₂ $-$ 2 C. Write the chemical formula for the following compounds a) Aluminium sulphate a) Aluminium sulphate $-$ Al ₂ SO ₁ b) Silver nitrate $AgNO_3$ c) Magnesium oxide $-$ MgO ₂ 3. Write the skeleton equation for the following word equation and then balance them a) Carbon + Oxygen \rightarrow Carbon dioxide b) Phophorus + Chloride. a) Carbon + Oxygen \rightarrow Carbon dioxide b) Phophorus + Chloride. a) Carbon + Oxygen \rightarrow Carbon dioxide b) Phophorus + Chloride. b) Carbon + Oxygen \rightarrow Carbon dioxide b) Phophorus + Chloride. b) Phophorus + Chloride. b) Phophorus + Chloride. b) Phophorus + Chloride. b) Carbon + Oxygen \rightarrow Carbon dioxide b) Phophorus + Chloride. a) Carbon + Oxygen \rightarrow Carbon dioxide b) Phophorus + Chloride. b) Carbon + Oxygen \rightarrow Carbon dioxide b) Phophorus + Chloride. b) Carbon + Oxygen \rightarrow Carbon dioxide b) Phophorus + Chloride. b) Carbon + Oxygen \rightarrow Carbon dioxide b) Phophorus + Chloride. b) Carbon + Oxygen \rightarrow Carbon dioxide b) Phophorus + Chloride. b) Carbon + Oxygen \rightarrow Carbon dioxide b) Phophorus + Chloride. c) Carbon + Oxygen \rightarrow Carbon dioxide b)		CO_2	\rightarrow C	$+ O_2$	a) C -					
a) CO - Carbon monoxide b) N ₂ O - Nitrogen oxide c) NO ₂ - Nitrogen di oxide d) Mg + HCl ₂ \rightarrow MgCl ₂ + H 4. Balance the following chemical equat a) Na + O ₂ \rightarrow Na ₂ O b) Ca + N ₂ \rightarrow Ca ₃ N ₂ c) N ₂ + H ₂ \rightarrow NH3 1. Find the valency of the element which is underlined in the following formula. a) NaCl b) CO ₂ c) Al (PO ₄) d) Ba (NO ₃) ₂ e) CaCl ₂ Answer: a) NaCl $-$ 1 b) CO ₂ $-$ 2 2. Write the chemical formula for the following compounds a) Aluminium sulphate $-$ Al ₂ SO ₄ b) Silver nitrate $-$ Magnesium oxide d) Barium chloride Answer: a) Aluminium sulphate $-$ Al ₂ SO ₄ b) Silver nitrate $-$ MgO ₂ d) Barium chloride $-$ MgO ₂ 3. Write the skeleton equation for the following word equation and then balance them a) Carbon + Oxygen \rightarrow Carbon dioxide b) Phosphorus + Chlorine \rightarrow Phosphorus pentachloride. a) Carbon + Oxygen \rightarrow Carbon dioxide b) Phosphorus + Chlorine \rightarrow Phosphorus pentachloride. b) Carbon + Oxygen \rightarrow Carbon dioxide b) Phosphorus + Chlorine \rightarrow Phosphorus pentachloride. b) Carbon + Oxygen \rightarrow Carbon dioxide b) Phosphorus + Chlorine \rightarrow Phosphorus pentachloride. c) Carbon + Oxygen \rightarrow Carbon dioxide b) Phosphorus + Chlorine \rightarrow Phosphorus pentachloride. c) Carbon + Oxygen \rightarrow Carbon dioxide c) Phosphorus + Chlorine \rightarrow Phosphorus pentachloride. c) Carbon + Oxygen \rightarrow Carbon dioxide c) Phosphorus + Chlorine \rightarrow Phosphorus pentachloride. c) Carbon + Oxygen \rightarrow Carbon dioxide c) Phosphorus + Chlorine \rightarrow Phosphorus pentachloride. c) Carbon + Oxygen \rightarrow Carbon dioxide c) Phosphorus + Chlorine \rightarrow Phosphorus pentachloride. c) Carbon + Oxygen \rightarrow Carbon dioxide c) Phosphorus + Chlorine \rightarrow Phosphorus pentachloride. c) Carbon + Oxygen \rightarrow Carbon dioxide c) Carbo		PCl ₅	\rightarrow P	$+ Cl_5$	b) P -	ving compounds.	the followin	names of t	7. Write the	
b) N_2O - Nitrogen oxide c) NO_2 - Nitrogen di oxide d) PCl_5 - Phosphorus penta chloride V. Answer the following. 1. Find the valency of the element which is underlined in the following formula. a) $Na + O2 \rightarrow Na_2O$ b) $Ca + N_2 \rightarrow Ca_3N_2$ c) $N_2 + H_2 \rightarrow NH3$ d) $CaCO_3 + HCl \rightarrow CaCl_2 + CO_2 + H_2$ Answer: a) $NaCl \rightarrow 1$ b) $\underline{CO}_2 \ c) \underline{Al} (PO_4) \ d) \underline{Ba} (NO_3)_2 \ c) \underline{CaCl_2}$ Answer: a) $NaCl \rightarrow 1$ b) $\underline{CO}_2 \ c - 2$ c) $\underline{Al} (PO_4) \ - 1$ d) $\underline{Ba} (NO_3)_2 \ - 2$ c) $\underline{NaCl}_2 \ - 2$ c) $N_2 + H_2 \rightarrow NH3$ d) $CaCO_3 + HCl \rightarrow CaCl_2 + CO_2 + H_2$ a) $Na + O_2 \ - 2AIN3$ d) $CaCO_3 + 2HCl \rightarrow CaT_2 \ - 2NH3$ d) $CaCO_3 + 2HCl \rightarrow CaCl_2 + CO_2 + H_2$ e) $Pb(NO_3)_2 \ - 2C$ b) $3Ca + N_2 \ - 2Ca_3N_2$ c) $N_2 + 3H_2 \ - 2NH3$ d) $CaCO_3 + 2HCl \rightarrow CaCl_2 + CO_2 + H_2$ e) $Pb(NO_3)_2 \ - 2C$ c) $N_2 + 3H_2 \ - 2NH3$ d) $CaCO_3 + 2HCl \rightarrow CaCl_2 + CO_2 + H_2$ e) $Pb(NO_3)_2 \ - 2C$ b) $Scl + N_2 \ - Ca_3N_2$ c) $N_2 + 3H_2 \ - 2NH3$ d) $CaCO_3 + 2HCl \rightarrow CaCl_2 + CO_2 + H_2$ e) $Pb(NO_3)_2 \ - 2C$ c) $Nager + 3H_2 \ - 2NH3$ d) $CaCO_3 + 2HCl \rightarrow CaCl_2 + CO_2 + H_2$ e) $Pb(NO_3)_2 \ - 2C$ lesson 05 MOVEMENT 1. Answer verv briefly. 1. Answer verv briefly. 1. What is skeleton? it is composed of connective tissue bones, cartilage, tendons and ligan 2. What is cranium? b) Silver nitrate a) Carbon + Oxygen \Rightarrow Carbon dioxide b) Phosphorus + Chlorine \Rightarrow Phosphorus pentachloride. BaCl ₂ 3. Write the skeleton equation for the following word equation and then balance them a) Carbon + Oxygen \Rightarrow Carbon dioxide b) Phosphorus + Chlorine \Rightarrow Phosphorus pentachloride. BaCl ₂ 3. Why our backbone is slightly moveable b) The discs of cartilage allow the verton move slightly to forward and do b) The discs of cartilage allow the verton move slightly to forward and do b) The discs of cartilage allow the verton move slightly to forward and do b) The discs of cartilage allow the verton move slightly to forward and do b) The discs of cartilage allow the verton move s		SO_2	\rightarrow S	$+ O_2$	c) S -		2 d) PCl5	2O c) NO2	a) CO b) N2	
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1. Find the valency of the element which is underlined in the following formula.d) $CaCO_3 + HCl \rightarrow CaCl_2 + CO_2 + HCl + CaCl_2 + CAC$		a_3N_2	$\rightarrow Ca_3N_2$	1 ₂	b) Ca + N	a chloride	horus penta c	- Phosph	d) PCl ₅	
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a) NaCl b) \underline{CO}_2 c) \underline{AI} (PO ₄) d) \underline{Ba} (NO ₃) ₂ e) $\underline{CaCl_2}$ Answer: a) Na Cl - 1 b) \underline{CO}_2 - 2 c) \underline{AI} (PO ₄) - 1 d) \underline{Ba} (NO ₃) ₂ - 2 c) $\underline{CaCl_2}$ - 2 2. Write the chemical formula for the following compounds a) Aluminium sulphate b) Silver nitrate c) Magnesium oxide d) Barium chloride Answer: a) Aluminium sulphate - Al ₂ SO ₄ b) Silver nitrate - MagnO ₃ c) Magnesium oxide - MgO ₂ d) Barium chloride - BaCl ₂ 3. Write the skeleton equation for the following word equation and then balance them a) Carbon + Oxygen \rightarrow Carbon dioxide b) Phosphorus + Chlorin \Rightarrow Phosphorus pentachloride.	⊦H ₂ O	$CaCl_2 + CO_2 + H_2$	$\rightarrow CaCl_2$	3 +HCl	d) CaCO ₃	nt which is	the element	valency of t	1. Find the	
Answer:a) \underline{NaCl} a) $Na + O_2 \rightarrow Na2O$ a) \underline{NaCl} -1b) $\underline{CO_2}$ -2c) \underline{Al} (PO ₄)-1d) \underline{Ba} (NO ₃) ₂ -2e) $\underline{CaCl_2}$ -2c) $\underline{Magnesium oxide}$ of $\underline{Nagnesium oxide}$ of $\underline{Nagno_3}$ 3a) Aluminium sulphate- Al_2SO_4 b) Silver nitrateAgNO_3c) Magnesium oxide-d) Barium chloride-3. Write the skeleton equation for the following word equation and then balance thema) Carbon + Oxygen \rightarrow Carbon dioxideb) Phosphorus + Chlorine \Rightarrow Phosphorus pentachloride.	O_2	$PbO + NO_2 + O_2$	\rightarrow PbO -	3)2	e) Pb(NO3	nula.	owing formu	in the follo	underlined	
a) \underline{NaCl} - 1 b) $\underline{CO_2}$ - 2 c) $\underline{Al} (PO_4)$ - 1 d) $\underline{Ba} (NO_3)_2$ - 2 c) $\underline{CaCl_2}$ - 2 c				R:	ANSWEF	$(\mathrm{NO}_3)_2 \mathrm{e}) \underline{\mathrm{Ca}} \mathrm{Cl}_2$	PO ₄) d) <u>Ba</u> (1	$\underline{CO}_2 c) \underline{Al} (P$	a) <u>N</u> aCl b) <u>(</u>	
b) $\underline{CO_2}$ - 2 c) \underline{AI} (PO ₄) - 1 d) \underline{Ba} (NO ₃) ₂ - 2 e) $\underline{CaCl_2}$ - 2 c. Write the chemical formula for the following compounds a) Aluminium sulphate b) Silver nitrate c) Magnesium oxide d) Barium chloride Answer: a) Aluminium sulphate - Al ₂ SO ₄ b) Silver nitrate AgNO ₃ c) Magnesium oxide - MgO ₂ d) Barium chloride - BaCl ₂ 3. Write the skeleton equation for the following word equation and then balance them a) Carbon + Oxygen \Rightarrow Carbon dioxide b) Phosphorus + Chlorine \Rightarrow Phosphorus pentachloride.		Ja ₂ O	\rightarrow Na ₂ O	\mathbf{D}_2	a) Na + O ₂				Answer:	
c) \underline{Al} (PO ₄) - 1 d) \underline{Ba} (NO ₃) ₂ - 2 e) $\underline{CaCl_2}$ - 2 c. Write the chemical formula for the following compounds a) Aluminium sulphate b) Silver nitrate c) Magnesium oxide d) Barium chloride Answer: a) Aluminium sulphate - Al ₂ SO ₄ b) Silver nitrate AgNO ₃ c) Magnesium oxide - MgO ₂ d) $CaCO_3 + 2HCI \rightarrow CaCl_2 + CO_2 + H e) Pb(NO3)_2 \rightarrow PbO + 2NO_2 + CO_3$ LESSON 05 MOVEMEN LAnswer very briefly. 1. What is skeleton? 1. What is skeleton? 1. What is composed of connective tissue bones, cartilage, tendons and ligar 2. What is cranium? b) Silver nitrate AgNO ₃ c) Magnesium oxide - MgO ₂ d) Barium chloride BaCl ₂ 3. Write the skeleton equation for the following word equation and then balance them a) Carbon + Oxygen \Rightarrow Carbon dioxide b) Phosphorus + Chlorine \Rightarrow Phosphorus pentachloride.				N ₂	b) 3Ca + N		- 1	-	a) <u>N</u> aCl	
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a) Culmhum (Ommenn). Culmhum diamide							•	-	a) C. 1. 1.	
c) Sulphur + Oxygen → Sulphur dioxide d) Magnesium + hydrogen → Magnesium + Hydrogen chloride chloride						Hydrogen	Magnesium + Hyd	- hydrogen \rightarrow Ma	-	

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4. Differentiate axial and appendicular b) Endoskeleton and Exoskleton Endoskeleton Exoskleton skeleton. It is found in the inside \blacktriangleright The axial skeleton consists of the skull. It is found in the facial bones, sternum, ribs, and vertebral of the human body exterior part of the body column. c) Pectoral and Pelvic girdle > The appendicular skeleton comprises the Pectoral girdle Pelvic girdle shoulder girdle; the arm, wrist, and hand Attachment of arms. Attachment of legs. bones; the pelvic girdle; and the leg, ankle, This forms a ball and This forms a ball and and foot bones. socket joint. socket joint. 5. What is ligament? d) Ball and socket Joint and Hinge Joint Bands of tough, elastic connective tissue Ball and socket Joint Hinge Joint that surround a joint to give support and A cylindrical protrusion A ball shaped head of limit the joint's movement. one bone articulates of one bone articulates 6. Define Muscle. with a trough-shaped with a cup like socket The muscles in the body provide the means of an adjacent bone. depression of an of all movements. adjacent bone. > They cover the skeletal framework and e) Voluntary and Involuntary muscle also give shape to the body Voluntary muscle Involuntary muscle 7. Differentiate tendons and ligament. Attached to bones. Attached to soft parts of Tendons Ligament Found in arms, legs, the body like blood fibrous cords of tissue fibrous cords of tissue neck. vessels, iris, bronchi that attach bone to that attach muscle to and the skin bone. 2. What are antagonistic muscles? Give one example. **II.** Answer briefly. Muscles often work in pairs which work 1. Differentiate between the following. against each other. These are called a) Movement and Locomotion. antagonistic pairs. Locomotion Movement ➤ The two muscles, the biceps and triceps are Movement is the act The movement of an working against each other. of changing the place organism from one When the biceps contracts the lower arm is or position by one or place to another is raised and the arm bends. In this position more parts of the body known as locomotion.

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bone

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the triceps muscle is relaxed.

3. How is the skeleton of a bird well-suited for flying?

- A bird has streamlined body. Its bones are light and strong.
- They are hollow and have air spaces between them.
- The breast bones are modified to hold massive flight muscles which help in moving wings up and down.
- Birds have special flight muscles and the forelimbs are modified as wings.
- The wings and tail have long feathers, which help in flying

4. What are the functions of skeleton in human body?

- > It provides structure and shape to the body.
- It supports and surrounds the internal organs of the body.
- Calcium and phosphorus, the two minerals that the body needs for important regulatory functions, are stored inside the bones.
- Red blood cells are produced in the bone marrow.
- The bones of the skeletal system act as levers for muscular action.

LESSON 06 REACHING THE

AGE OF ADOLESCENCE

- I. Answer briefly.
- 1. What is adolescence?
 - The term adolescence is derived from the Latin word 'adolescere' meaning 'to grow' or 'grow to maturity'

2. List out the changes which occur during

- puberty.
 - Changes in body size
 - Changes in body proportion
 - Development of primary sex characteristics
 - Development of secondary sex characteristics
- 3. What is secondary sex characteristics?
 - > Hair
 - > Skin
 - Glands
 - Muscles
 - ➢ Voice

4. What is fertilization?

The action of fertilizing an egg or a female animal or plant, involving the fusion of male and female gametes to form a zygote.

5. Explain Menarche.

- The first menstrual flow begins at puberty and is termed menarche.
- The ova begin to mature with the onset of puberty.

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6. Explain the process of pregnancy.

- After ovulation the ovum reaches the fallopian tube and fertilization takes place.
- The fertilized egg undergoes development and it is implanted in the uterus.
- The corpus luteum continues to grow and produces large amount of progesterone.
 This results in pregnancy

7. Explain the importance of cleanliness during the time of menstrual cycle in girls.

- ➢ Have a bath once or twice a day.
- Change the underwear daily.

8. How is adolescence differ from childhood?

- During childhood, legs grow proportionately more than the trunk.
- At the time of puberty trunk also lengthens. Also, trunk broadens at the hip and shoulder thus giving the adult proportion to the body

II. Answer in detail.

1. What are the physical changes that occur in boys and girls during adolescence?

Girls	Boys
Height and weight	Height and weight increase.
increase.	
Hip broadens.	Shoulder broadens.
Hair grows in arm	Hair grows in the arm pits and
pits and pubic	pubic area, and facial hair also
area.	appears.
Voice becomes	Voice break takes place due
shrill.	to lengthening of vocal cord
	and enlarging of larynx.

2. Explain the role of hormones in

reproduction.

Follicle Stimulating Hormone (FSH):

In the male it is necessary for the development of seminiferous tubules, and for spermatogenesis.

Luteinizing Hormone (LH):

In the female, it is the hormone necessary for ovulation.

Prolactin (PRL) or Lactogenic Hormone:

The main function of this hormone is milk secretion during lactation.

Oxytocin Hormone:

It is involved in the contraction of smooth muscles of uterus during child birth.

3. Briefly describe the menstrual cycle.

- Usually one mature ovum (or egg) is released from one of the ovaries into the oviduct once in every 28 days. This is called ovulation.
- Before ovulation (or release of ovum), the inner wall of uterus becomes thick and spongy, and full of tiny blood vessels (or blood capillaries). It prepares itself to receive the fertilized ovum.
- If the ovum does not get fertilized then the thick and soft inner lining of uterus is no longer needed and hence it breaks. So, the thick and soft inner lining of uterus along with the blood vessels and the dead ovum comes out of the vagina in the form of a bleeding called menstruation.

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- Menstruation usually occurs 14 days after ovulation and usually lasts for about 3 to 5 days.
- After menstruation is over, the inner lining of the uterus starts building up again so that it may become ready to receive the next ovum.

4. Briefly explain the nutritional needs of adolescence.

Minerals

Since there is an increase in skeletal mass and blood volume during adolescence, the body needs calcium, phosphorus and iron.

Calcium

Calcium intake needs to be increased to prevent osteoporosis in later life. It is present in milk and milk products or other equivalents.

Iodine

It helps to prevent thyroid gland related diseases.

Iron

Iron builds blood and iron-rich foods such as green leafy vegetables, jaggery, meat, dates, fish, chicken, citrus, etc.

LESSON 07 DIGITAL

PAINTING

I. Answer briefly.

- 1. What is Tux Paint?
 - Tux Paint is a free drawing program designed for young children.
- 2. What is the use of Text Tool?
 - \succ This tool is used to type texts.
- 3. What is the shortcut key for Save option?
 - \succ Ctrl + S

4. What is Tux Math?

Tux Math is an open source arcade – style video game for learning arithmetic. The main goal is to make learning effective and fun.

5. What is the use of Ranger ?

► Ranger - addition, subtraction,

multiplication and division to ten.

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