#### **REVISION EXAM JAN-2023**

### XII<sup>th</sup> - MODEL EXAM

## Time Allowed: 1.30Hours] [Maximum Marks: 50 CHEMISTRY PART – I

Answer any *Ten* of the following questions. 10x2=20

- 1. What is Smelting?
- 2. What is Auto reduction?
- 3. Write the uses of potassium dichromate.
- 4. Explain why  $Mn^{2+}$  is more stable than  $Mn^{3+}$ ?
- 5. What is crystal field splitting energy?
- 6. What is Bragg's equation?
- 7. Give two examples of zero order reaction.
- 8. Calculate the pH of 0.001M HCl solution.
- 9. State kohlraush's law.
- 10. What are buffer solutions?
- 11. What is meant by salt hydrolysis?
- 12. Write the Zwitter ion structure of alanine
- 13. What are food preservatives?
- 14. Write a short note on peptide bond
- 15. Write a note on vulcanization of rubber?

#### PART - II

Answer any Ten of the following questions.

10x3=30

- 16. What is the role of Limestone in the extraction of Iron from itoxideFe<sub>2</sub>O<sub>3</sub>?
- 17. What are the difference between minerals and ores?
- 18. Describe the preparation of potassium dichromate.
- 19. Why transition elements form complexes.
- 20. Give the difference between double salts and coordination compounds.
- 21. What is meant by packing efficiency?
- 22. Distinguish: rate of a reaction and rate constant of a reaction?
- 23. State: Ostwald's dilution law.
- 24. State Faraday's Laws of electrolysis
- 25. Write a note on ultrafiltration.
- 26. Define: Buffer index (β).
- 27. How are vitamins classified
- 28. What are drugs? How are they classified
- 29. Give the three differences between DNA and RNA.

# 30. How is Teflon prepared?

