UNIT TEST-11A (HYDROXY COMPOUNDS AND ETHERS)

CLASS: XII MARK: 35 SUB : CHEMISTRY (page no 104 to 122) **TIME: 1.30 HRS `PART-I** I. Choose and write the correct answer: 5 X 1 = 51. Which one of the following is the strongest acid b) 4 – chlorophenol a) 2 - nitrophenol c) 4 – nitrophenol d) 3 – nitrophenol 2. Which of the following compounds on reaction with methyl magnesium bromide will give tertiary alcohol. a) benzaldehyde b) propanoic acid c) methyl propanoate d) acetaldehyde 3. Ketone react with Grignard reagent to give ___ a) primary alcohol b) secondary alcohol c) tertiary alcohol d) none of these 4. Secondary alcohol do not react with KOH because a) no alpha hydrogen b) no beta hydrogen c) no gamma hydrogen d) none of these 5. Which one of the following is used for sweetening agent a) methanol b) ethanol c) glycol d) glycerol **PART-II** II. Answer any three questions (q.no.10 is compulsory) $3 \times 2 = 6$ 6. Explain – Hydroboration 7. How will you prepare butan-2-ol from Grignard reagent? 8. How are Convert glycerol to acrolein? 9. What are the uses of glycerol? 10 How is crotyl alcohol obtained by reduction reaction? **PART-III** III. Answer any three questions $3 \times 3 = 9$ 11. Explain Swern oxidation 12. Distinguish between 1⁰, 2⁰ and 3⁰ alcohols using Luca's reagent. 13. How are the nitroglycerin prepared from glycerol? 14. . how will you prepare the following by using grignard reagent? a) Propan-1-ol b)propan-2-ol 15. Write note biological oxidation? **PART-IV** IV. Answer any three questions $3 \times 5 = 15$ 16. Write a note on Sayzteff's rule? 17. Distinguish between 1^o, 2^o and 3^o alcohols using Victor Meyer's test 18. i) What is Baeyer's regent? how it is useful to convert ethene to ethane 1 2 diol? ii) How is ethylene glycol converted into 1,4 dioxane? 19. How are the following conversion effected? i) ethylene glycol → acetaldehyde ii) glycerol → acrolein 20. i) How are the glycerose prepared from glycerol? ii) Write the glycol react with following reagent i) dil H₂SO₄ ii) Con H₂SO₄
