CHEMISTRY	
STD: 12	Marks : 70
Time: 3.00 hrs	15 x 1 = 15
I. Choose the correct answer:	
1 Zinc is obtained from ZnO by a) Carbon reduction b) Reduction using silver 2 Oxidation state of carbon in its hydrides 3 Hybridization of XeOF4 is a) sp <sup>3</sup> b) sp <sup>3</sup> d	etrochemical process d) Acid leaching
2. Oxidation state of carbon in its hydrocal	sp'd2 d) sp'd1
3. Hybridization of XeOF4 is a) sp	tric analysis.
<ul> <li>Assertion: Ce<sup>4*</sup> is used as an oxidizing agent in volume Reason: Ce<sup>4*</sup> has the tendency of attaining +3 oxidate.</li> <li>Both assertion and reason are true and reason is the combined by Both assertion and reason are true but reason is not the combined part of th</li></ul>	tion state.
5. Primary and secondary valencies of Ni in [Ni(en) <sub>3</sub> ]Cl <sub>2</sub> a) 3, 3 b) 3, 2 2, 6 d) 6, 2	
6. The ratio of close packed atoms to tetrahedral hole is a) 1:1 b) 1:2 c) 2:1 d) 1:4	
7. If 50% of a first order reaction is completed in 60 minute complete is a) 50 min b) 60 min c) 90 min d) 120	s, 75% of the same reaction would ) min
8. Which of the following can act as Lowry – Bronsted acid a) HCl b) SO <sub>4</sub> <sup>2</sup> c) HPO <sub>4</sub> <sup>2</sup> d) Br	as well as base?
is used in Laptop     a) Lead storage battery b) Fuel cell c) Mercury buttor	cell d) Lithium in battery
10. Collodion is a 4% solution of which one of the following	ng compounds in alcohol – ether mixture? coldinitrate d) Nitrocellulose
11. Carbolic acid is a Phenol b) Picric acid c) benzoi	ic acid d) phenylacetic acid
12. The reagent used to distinguish between acetaldehyde a) Tollens reagent b) Fehling's solution c) 2,4	and benzaldehyde is  – dinitrophenyl hydrazine d) semicarbazide
<ol> <li>The basic character of amine is due to the</li> <li>a) tetrahedral structure</li> <li>b) presence of nitrogen atom</li> <li>d) high electronegativity of nitrogen.</li> </ol>	c) lone pair of electrons on nitrogen atom
14. Which one of the following is not produced by body? <ul> <li>a) DNA</li> <li>b) Enzymes</li> <li>c) Harmones</li> </ul>	Vitamins
15. Non stick cook wares generally have a coating of a po a) ethane b) prop-2-enenitrile c) chloroethene PART -	d) 1,1,2,2-tetrafluoroethane
Answer any SIX questions. (Q. No 24 is Compu	
16. Write note on Auto – reduction?	0 X Z= 12
17. What is innert pair effect?	
Kindly send me your answer keys to us	s - padasalai net@omail.com

PRIT~EDUCATION;SS. PRITHIVIGGIIGKNEOR MORE www.TrbTnpsc.com
20. Why are lyophillic colloidal sols are more stable than lyophobic colloidal sol?
21 Write Trans esterification reaction.
22. Aniline does not undergo Friedel - Crafts reaction why? 23. Name the Vitamins whose deficiency cause i) Cheilosis ii) Depression
23. Name the Vitamins whose deficiency cause i) Cheilosis ii) Depression
24. Calculate the Magnetic moment and magnetic property of [Fe(CN)6]3-
PART - III
Answer any SIX questions. (Q. No 33 is Compulsory)
25. Explain the following terms with suitable example. i) Gangue ii) slag
26. Give any three uses of Borax.
27. State Hume – Rothery rule for alloy formation.
<ol> <li>Aluminium crystallizes in a cubic close packed structure. Its metallic radius is 125pm.</li> </ol>
calculate the edge length of unit cell.
29. What is an elementary reaction? Give the differences between order and molecularity
of a reaction.
30. Explain Kolbe's reaction?
31. Write the mechanism of Cannizarro reaction.
32. How is prepared Decron? Give its any one use?
33. The conductivity of a 0.01M solution of a 1:1 weak electrolyte at 298K is 1.5 x 10 <sup>-4</sup> S cm <sup>-1</sup> .
Calculate molar conductivity of the solution
PART - IV
Answer all the questions.
34. a) i) Give the basic requirement for vapour phase refining (2)
ii) Describe a method for fefining Nickel. (3) (or)
b) i) Write note on Fisher tropsch synthesis? (2)
ii) Give the balanced equation for the reaction between the
ii) Give the balanced equation for the reaction between chlorine with cold NaOH and
hot NaOH. (3)
35. a) i) Which is more stable? Fe <sup>3+</sup> or Fe <sup>2+</sup> explain. (2)
ii) Explain Chromyl chloride test ? (3)
(or)
b) Write the postulates of Werner's theory. (5)
26 a) i) Distinguish between tetrahedral and a state to the same
36. a) i) Distinguish between tetrahedral and octahedral voids? (Any Two) (2)
ii) Explain: Schottky defect (3)
(or)
b) Derive Henderson equation (5)
37. a) i) Why does conductivity of a solution decrease on dilution of the solution ? (2)
ii) Explain Tyndall effect (3)
(or)
b) i) What is Urotropine? Give the structure? (2)
ii) Write a note on Levine and Hauser acetylation reaction. (3)
8 a) Compound (A) molecular formula C6H6O gives purple colouration with neutral FeCl3 Compound
(A) reacts with CHCl <sub>3</sub> / NaOH gives compound (B) and compound (A) also reacts with benzene
diazonium chloride in presence of NaOH at 273 – 278K gives compound (C). Identify the
compounds A. B. C and write the masting (6)
compounds A, B, C and write the reaction. (5)
(or)
b) i) What are harmones? Give an example (2)
ii) Write note on Buna – N rupper? (3)
The state of the s