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SRI RAMANA MAHARSH I MATRICULATION HIGHER SECONDARY SCHOOL

KAVERYAMPOONDI, K.Rajkumar 9843239843 Tiruvannamalai

SUBJECT: SCIENCE		Unit:	Unit: 1&7 (Model Test)				MARKS: 75	
I.CHOOSE THE CORRECT ANSWER:								20 X 1 =20
1.Inertia of a body depends on (a)weight of the object (bacceleration due to gravity of the planet (c)mass of the object								
2 Newton's III law is applicable								
a) for a body is at rest (b) for a body in motion (c) both a 8 b								
(d)Only for bodies with equal masses								
3.In which of the following sport the turning of effect of force used								
(a)Swimming (b)tennis (c)Cycling (d)hockey								
4.The unit of 'g	' is ms ⁻² .lt ca	n be also expre	ssed as		(0)11001			
(a)cm ^{s-1}	(b)N k	(g ⁻¹	(c)N m	² kg ⁻¹			(d)cm ² s ⁻²	
5.One kilogram	force equals	to	(-)	0	()		(-)	
(a)9.8 dyne	(b)9.8x10 ⁴ N	(c)98x10 ⁴ dyn	e	(d)980	dyne			
6.SI unit of imp	ulse is	., ,						
a)Ns	b)Ns ²		c)kg m	s ⁻²			d)kgm ² s ⁻²	
7. The gravitational force of earth acting on a body of mass 1kg is								
a)8.9N	b)9.8N	c)980		d)iN				
8.First systematic study of gravity								
a)Newton		b)Einsten		c)Galile	90			
9.Gravity of mo	oon a)9.8	b)8.9		c)1.625	5			
10.Example of Newton third law								
a)bird fly	b)Mo	mentum	c)Inert	ia				
11.Mass of 1 m	ole of Nitrog	en atom is						
a)28 amu		b)14 amu		c)28 g		d)14g		
12.The Volume occupied by 1 mole of a diatomic gas at S.T.P is								
a)11.2litre	b)5.6	litre	c)22.4	litre		d)44.8	itre	
13.Which of the following has the smallest mass?								
a)6.023 x 10 ²³ a	atoms of He	b)1 atom of ⊦	le	c)2g of	He	d)1 mo	le atoms of H	le
14.Which of the following is a triatomic molecule?								
a)Glucose	b)Heli	ium	c)Carb	on dioxi	de	d)hydro	ogen	
15.In the nucles of 20Ca40 ,there are								
a)20 protons and 40 neutrons b)20 protons and20neutron c)10protons and 40								
electrons d)40 protons and20 electrons								
16. Molar mass of CO2 a. 40g b. 42g c. 44g 17 Gram malagular mass of UCL a. 25g b. 26g c. 26 5g								
17. Gram mole kindly 18. The atom w	cular mass of send me you vas proposed	HCI a. 35g b r key Answers by a. Rob	o. 36g (c. to our pert	36.5g email id b. Dalto	l - <mark>pada</mark> on	salai.ne c. C.V.R	e <mark>t@gmail.co</mark> aman	m

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- 19. The mass of an atom can be converted into
- 20. One mole of matter contains a. 6.023

II. FILL IN THE BLANKS

- 1. To produce a displacement is required
- 2. is used to change the speed of car.
- 3. A man of mass 100 kg has a weight of at the surface of the Earth.
- 4. The value of G=
- 5. Radius of earth
- 6. Mass of the Earth
- 7. Lift is falling down freely R=
- 8. The average atomic mass of hydrogen is amu.
- 9. The number of atoms present in a molecule is called its
- 10. Atomicity of phosphorous is

III. MATCH THE FOLLOWING

- 1. Newton 1st law Flying bird
- 2. Newton 2nd law Stable equilibrium
- 3. Newton 3rd law Law of force
- 4. Gravity of Earth R>W
- 5. Lift upward 9.8ms⁻²

IV. Answer the following (any 10)

- 1. Define inertia. Give its classification.
- 2. Differentiate mass and weight.
- 3. Define moment of a couple.
- 4. State Newton's second law.
- 5. Why a spanner with a long handle is preferred to tighten screws in heavy vehicles?
- 6. State Newton's third law.
- 7. Define: Relative atomic mass.
- 8. Define: Atomicity.
- 9. Give any two examples for heterodiatomic molecules.
- 10. Find the percentage of nitrogen in ammonia.
- 11. Define isotopes.
- 12. Applications of Avogadro's law any two points.

V. Answer briefly.

- 1. State the universal law of gravitation and derive its mathematical expression.
- 2. Give the slient features of "Modern atomic theory".
- 3. How many grams are there in the following?

i. 2 mols of H₂ ii. 3 mols of Cl₂ iii. 5 mols of S₈iv. 4 mols of P₄ By,

K.Rajkumar 9843239843 Tiruvannamalai

a. $a=MC^2$ B. $E=MC^2$ C. $D=MC^2$ b. 6.023x10-²³ C. 8.023x10-²³

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10X1=10

10x2=20

10X1=10

4 moles

98 g

2 moles

0.25 moles

6. 8g of O₂

8. 4g H₂

7.112 g of N₂

9. Volume of STP-

10. Mass of H₂so₄-

22.4Lit

3x5=15

By, K.Rajkumar 9843239843 Tiruvannamalai