## DIRECTORATE OF GOVERNMENT EXAMINATIONS S.S.L.C. PUBLIC EXAM- APRIL 2024 SCIENCE ANSWER KEY

Part - I

## **Answer all the Questions:**

 $12 \times 1 = 12$ 

1.	(b)	Stem	1
2.	(c)	Fatty matter	1
3.	(d)	8.31 J Mol <sup>-1</sup> K <sup>-1</sup>	1
4.	(c)	Electrical Energy	1
5.	(b)	Restriction endonucleus	1
6.	(a)	6.023×10 <sup>23</sup>	1
7.	(b)	Pituitary Gland	1
8.	(c)	The flowers are brightly coloured have smell and nectar	1
9.	(c)	Mass of the object	1
10.	(c)	Atrium → Ventricle → Arteries → Vein	1

11.	(c)	$2CO_2 + O_{2(g)} \rightarrow 2CO_{2(g)}$	1
12.	(c)	Carcinoma	1

Part – II

Answer any Six questions. Question No.24 is compulsory. 7 x 2 = 14

	Coefficient of apparent expansion:	
13	Coefficient of apparent expansion is defined as the ratio of	1 1/2
	the apparent rise in the volume of the liquid per degree rise	
	in temperature to it unit volume	1/2
	It's SI unit is K-1	11/2
	Tungsten has a very high melting point.	
14	If it is used in fuse wire, it will not melt when large	2
	current passes through it	
	The appliances will get damaged	
	Rust:	
15	Rust is brown coloured hydrated ferric oxide.	2
	❖ 4Fe + $3O_2$ + X.H <sub>2</sub> O → 2Fe <sub>2</sub> O <sub>3</sub> .XH <sub>2</sub> O	
	Stage:	
	Stage is the background appearing when we open	
16	the scratch window.	2
	The background will most often be white.	
	We can change the background colour as we like	
		1
17	SA node acts as the pacemaker of the heart.	
17	It is capable of initiating impulse which can stimulate	
	the heart muscles to contract	1
		-
	Parts of hind brain:	
18	❖ Cerebellum	2
	❖ Pons	
	Medulla Oblangata	

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	Given:	
	P <sup>H</sup> = 4.5	
	P <sup>OH</sup> = ?	1
22		
	Solution :	
	$P^{H} + P^{OH} = 14$	
	$P^{OH} = 14 - 4.5$	1
	P <sup>OH</sup> = 9.5	

Part - III

Answer any Seven questions. Question No.32 is compulsory. 7x 4 = 28

23	Types of Inertia:		1
	Inertia of rest		
	Inertia of motion		
	Inertia of direction		
	a) Inertia of rest :		
	To resist a body to char	nge its state of rest.	
	Ex: After shaking leaves	s fall down.	
	b) Inertia of motion :	O'	
	To resist a body to char	nge its state of motion.	
	Ex: An athlete runs som	ne distance before jumping.	
	c) Inertia of direction :		
	To resist a body to char	nge its direction.	
	Ex : A sharp turn while of	driving a car you tend to lean side	
	way.		
24	a)		
	Natural Radioactivity	Artificial Radioactivity	
	❖ It cannot be controlled	❖ It can be controlled	1
	<ul> <li>Spontaneous process</li> </ul>	❖ Induced process	
	Alpha, Beta and gamma radiations are emitted	<ul> <li>Neutron,Positrons are emitted</li> </ul>	1
	Tadiations are emitted	emitted	
	b) Electric Heater, Electric Iron (Iro	on Box)	
	,,	,	1
1			

		1
25	a) When magnesium sulphate heptahydrate crystals are gently heated, it loses seven water molecules, and becomes anhydrous magnesium sulphate  Heating  MgSO4 . 7H2 O  Cooling (Magnesium Sulphate Magnesium heptahydrate)  B) Solubility is defined as the number of grams of a solute that can be dissolved in 100 g of a solvent to form its saturated solution at a given temperature and pressure	2
26	a) RQ = Volume of CO <sub>2</sub> liberated	
20	Volume of O₂ consumed  b)  ♣ During light independent reaction, CO₂ is reduced into carbohydrates with the help of ATP and NADPH₂  ♣ So light dependent reaction occur before the light independent reaction.	2
	Dental formula of rabbit :	2
27	= 2	

	1	
	C = 0	
	0	1
	PM = 3	
	$\overline{2}$	
	M = 3	1
	3	1
		1
28	a)	
	Euploid considered to be advantageous to both plants and	2
	animals, as they often result in increase fruit and flower size.	
	b) i) Unipolar neuron:	
	Only one nerve process arises from the cyton.	
	ii) Bipolar neuron:	
	<ul> <li>Cyton gives rise to two nerve processes</li> </ul>	
	iii) Multipolar neuron :	
	❖ The cyton gives rise to many dendrons and an axon found in	2
	cerebral cortex of brain.	

* Distributing vessels	4
❖ Deep location       ❖ Superficial in location         ❖ Blood flow with high pressure       ❖ Blood flow with low pressure         ❖ Wall of artery is strong thick and elastic       ❖ Wall of vein is weak thin and non-elastic         ❖ All arteries carry oxygenated blood except pulmonary arteries       ❖ All veins carry deoxygenated blood except pulmonary veins         30       Ethnobotany:         Ethnobotany is the study of regions plants and their practical uses through the traditional knowledge of the local culture of people.         Importance :       ❖ It provides traditional uses of plant.	4
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	2
❖ It gives information about certain unknown and known useful	
plants.	
a) Consequences of deforestation : (Any 4 points)	
<ul> <li>❖ Flood</li> <li>❖ Drought</li> <li>❖ Soil erosion</li> <li>❖ Loss of wild life</li> </ul>	2
Extinction of species	
❖ Imbalance of biogeochemical cycles	
<ul> <li>Alteration of climate condition.</li> <li>Desertification</li> </ul>	
b) Applications of DNA finger printing technique: (Any 2 points)  DNA finger printing technique is widely used in forensic	
applications like crime investigation such as identifying the culprit  It is used in paternity testing incase of disputes.	2

	It helps in the study of genetic diversity of population, evolution and speciation.	
32	a) 1. The acid that renders aluminium passive is dilute or	1
	concentrated nitric acid.	
	2. Aluminium becomes passive due to the formation of an	
	oxide film on its surface.	1
	b) Number of moles = Number of molecules of NH <sub>4</sub> Cl	-
	Avagadro Number	
	$= 1.51 \times 10^{23}$	
	$\overline{6.023 \times 10^{23}}$	
	= 1/4	
	= 0.25 moles of NH₄Cl	
		1

## Part - IV

## Answer all the question:

 $3 \times 7 = 21$ 

33	a) (Any 2 points)
	i)
	❖ Convex lens is used in camera lenses and magnifying
	lenses.
	Used in making microscope, telescope and slide projectors.
	Used to correct the object of vision called hyper metropia.
	ii)
	❖ When a beam of white light or composite light is refracted

through any transparent modia such as glass or water it	
through any transparent media such as glass or water, it	
splits into its component colours.	
This phenomenon is called as dispersion of light.	2
iii)	
❖ As the red light has highest wavelength among all the	
colours, it is scattered least.	
❖ It travels a longer distance in the atmosphere.	
iv) Least count of travelling microscope : 0.01 mm	
iv) Least count of travelling Thicroscope . 0.01 min	<b>&gt;</b>
	2
	1
b)	
i) Echo:	
An Echo is the sound reproduced due to the reflection of the	
original sound from various rigid surfaces.	1
ii)	-
Minimum time gap between the original sound and an echo must be 0.1 s.	
❖ Minimum distance required to hear an echo is 17.2 m.	1
iii)	1
<ul> <li>❖ Safe testing tool.</li> </ul>	
	1
iv) Speed of sound = Distance travelled	
Time taken	1
= 2d / t	-
	1

		2
34	i) Number of Moles of O <sub>2</sub> = Volume of S.T.P	
(a)	Molar Volume = 3 / 22.4	>
	= 0.1339 moles	
	Number of Molecules = Number of moles x Avagadro number	
	$= 0.1339 \times 6.023 \times 10^{23}$	
	$= 0.8064 \times 10^{23}$	
	= 8.064 x 10 <sup>22</sup> O <sub>2</sub> molecules	
	Number of moles of $Cl_2 = 5 / 22.4 = 0.2232$ moles	
	Number of molecules = $0.2232 \times 6.023 \times 10^{23}$	
	= 1.344 x 1023 molecules	
	Number of moles of $H_2$ = 6 / 22.4 = 0.2678 moles	
	Number of molecules = $0.2678 \times 6.023 \times 10^{23}$	
	= 1.6129 x 10 <sup>23</sup> molecules	
	1) 6 litre of H <sub>2</sub> has the highest number of molecules	
	2) 3 litre of O <sub>2</sub> has the lowest number of molecules	
	ii)	
	❖ An atom is no longer indivisible.	
	❖ Atoms of the same element may have different atomic mass.	
	❖ Atoms of different element can be transmitted into atoms of	
	other elements	
	❖ Atom is no longer indestructive.	
	❖ Atoms may not always combine in a simple whole number	
	ratio.	

❖ Atom is the smallest particle that takes part in a chemical				
reaction.				
❖ The mass of an atom can be converted into energy (E=mc²)				
i)				
Some detergents having a branched hydro carbons chain				
are not fully biodegradable by micro-organisms present in				
water.	2			
❖ So they cause water pollution.	<b>-</b>			
ii)				
❖ A → Ethanoic acid				
CH₃ COOH				
❖ Esterification	2			
	2			
	1			
i) Synthetic auxin :				
❖ Artifically synthesized auxin that have properties like auxins				
are called synthetic auxins.				
Eg : 2-4-D	1			
ii) Structure of Ovule:				
❖ Nucleus is enclosed by two integuments leaving an opening				
called as micropyle.				
❖ The ovule is attached to ovary wall by a stalk known as				
funiculus.	1			
❖ Chalaza is the basal part	<b>1</b>			
❖ The embryo sac contains seven cells and the eighth nuclei				
located within the nucleus				
	reaction.  The mass of an atom can be converted into energy (E=mc²)  Some detergents having a branched hydro carbons chain are not fully biodegradable by micro-organisms present in water.  So they cause water pollution.  CH₃ COOH  CH₃ COOH  CH₃ COOH  CH₃ COOH  CH₃ COOC₂H₅ + H₂O  Esterification  i) Synthetic auxin:  Artifically synthesized auxin that have properties like auxins are called synthetic auxins.  Eg: 2-4-D  ii) Structure of Ovule:  Nucleus is enclosed by two integuments leaving an opening called as micropyle.  The ovule is attached to ovary wall by a stalk known as funiculus.  Chalaza is the basal part  The embryo sac contains seven cells and the eighth nuclei			

	❖ 'The three cells at the chalaza end are the antipodal cells.				
	I	3			
					2
35	i) Father of Indian Gre		ion:	7.0	
b)	Dr.M.S.Swaminathan ii)				
	Out breading		Inbreading	1	
	❖ Cross between			of closely related	
	different species with animals within the same breed desirable features of for about 4-6 generation				
	economic value are mated.				1
	❖ The hybrids are stronger It helps in the accumulation of				
	and vigorous than their superior gene parents undesirable ge			genes and eliminate le genes.	
	♣ Eg: Mute       Eg : Sheep Hissardale				
					1
3 <i>5</i> (b)	Factors	Тур	e -1	Type – 2	
(6)	Prevalence	10-20 %		80-90%	
	Age of onset Juvenile of		nset	Maturity onset	
	(< 20 year		rs)	(> 30 years)	
	Body weight	Normal ( weight	or) under	Obese	4

Defect	Insulin deficiency	Target cells do	
	due to destruction	respond to insulin	
	of $\beta$ cells		
Treatment	Insulin	Can be controlled	
	administration in	by diet, exercise	
	necessary	and medicine	

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