

## மருதம் அகாடமி Youtube channel, கணியூர்

தொகுப்பு: ந. சண்முகசுந்தரம் (மருதம் ஆசிரியர்), அ.எண்: 96598 38789

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### HIGHER SECONDARY SECOND YEAR

### 12<sup>th</sup> – CHEMISTRY / UNIT 1 - METALLURGY

#### I. Choose the correct answer:

10 X 1 = 10

- Roasting of sulphide ore gives the gas (A). (A) is a colourless gas. Aqueous solution of (A) is acidic. The gas (A) is  
a) CO<sub>2</sub>                      b) SO<sub>3</sub>                      c) SO<sub>2</sub>                      d) H<sub>2</sub>S
- Which of the metal is extracted by Hall-Heroult process?  
a) Al                      b) Ni                      c) Cu                      d) Zn
- Electrochemical process is used to extract  
a) Iron                      b) Lead                      c) Sodium                      d) silver
- Flux is a substance which is used to convert  
a) Mineral into silicate                      b) Infusible impurities to soluble impurities  
c) Soluble impurities to infusible impurities                      d) All of these
- Zinc is obtained from ZnO by  
a) Carbon reduction                      b) Reduction using silver  
c) Electrochemical process                      d) acid leaching
- Considering Ellingham diagram, which of the following metals can be used to reduce alumina?  
a) Fe                      b) Cu                      c) Mg                      d) Zn
- Which of the following is used for concentrating ore in metallurgy?  
a) Leaching                      b) Roasting                      c) Froth floatation                      d) Both (a) and (c)
- In the electrolytic refining of copper, which one of the following is used as anode?  
a) Pure copper                      b) Impure copper                      c) Carbon rod                      d) Platinum electrode
- Which one of the following reaction represents calcinations?  
a)  $2Zn + O_2 \rightarrow 2ZnO$                       b)  $2ZnS + 3O_2 \rightarrow ZnO + 2SO_2$   
c)  $MgCO_3 \rightarrow MgO + CO_2$                       d) Both (a) and (c)
- The metal oxide which cannot be reduced to metal by carbon is  
a) PbO                      b) Al<sub>2</sub>O<sub>3</sub>                      c) ZnO                      d) FeO

#### II Answer any eight questions:

8 X 2 = 16

- What is Liquefaction?
- What is Ellingham Diagram?
- Why Ag<sub>2</sub>O and HgO is Unstable?
- Give the Uses of Gold?
- Write the Uses of Copper?
- Write the Uses of Gold?
- What is Auto reduction?
- What is Gravity separation (or) Hydraulic wash?
- What is Acid leaching?
- What is Calcination?

#### III Answer any eight questions:

8 X 3 = 24

- What are the difference between minerals and ores?
- What is the role Limestone in the extraction of Iron from its Oxide Fe<sub>2</sub>O<sub>3</sub>?
- Suitable Examples: i) Gangue                      ii) Slag
- Write a note on alkali Leaching with an example.
- Give the Limitation of Ellingham diagram?
- Explain Mond's process.
- Describe the role of Iodine in the refining of Zirconium?
- Explain the principle of electrolytic refining with an example
- Write a short note on electrochemical principles of metallurgy.
- Give the basic requirement for vapour phase refining.

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10<sup>th</sup> to 12<sup>th</sup> important Questions.