UNIT TEST -I TIME : 1.00 hr CLASS : XII SUBJECT: CHEMISTRY (METALLURGY) MARKS: 40 PART-I **CHOOSE THE CORRECT ANSWER:** 10 X 1 = 10 1.wolframite ore is separated from tinstone by the process of _ (a) electromagnetic separation. (b) smelting (c) calcination (d) Roasting 2. the incorrect statement among the following is (a) Nickel is refined by mond's process (b) Titanium is refined by van-Arkel's process (c) Zinc blende (ZnS) is connected by froth floatation process (d) In the metallurgy of gold the metal is leached with dilute sodium chloride solution 3. Match items in column-I with the items of column-II and assign the correct code Column-I Column-II (A) Hydrogen reduction ZnO (i) (B) Carbon reduction (ii) HgS (C) Metal reduction (iii) AgCI (D) Auto reduction (iv) B_2O_3 (v) Fe_3O_4 (a) A-iii , B-v ,C-i , D-ii (b) A-iv , B-ii ,C -v , D -iii (d) A-v , B- i ,C -iv , D -ii (c) A-v, B-iii, C-i, D-ii 4. Extraction of gold involves leaching with cyanide ion Gold is later recovered by : (a) metal displacement with zinc. (b) Liquation (c) Distillation (d) Zone refining 5. The following set of reactions are used in refining zirconium. This method is called as Zrl₄ $Zr(impure) + 2I_2$ 1800K Zr(pure) + 2l₂ Zrl₄ (a) Zone refining (b) Liquation (c) Mond's process (d) van Arkel process 6.Bauxite has the composition: a)Al₂O₃.nH₂Ob) Fe₂O₃.2H₂O d) none of the above c) Al_2O_3 7.Zinc is obtained from ZnO by a) carbon reduction b) reduction using silver c) Electrochemical process d) Acid leaching 8. The process of conversion of gold complex into elemental gold is _____ (a) Cupellation (b) cementation (c) liquation (d) concentration

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9.Match: (1) Cyanide process (i) Ultrapure Ge (2) Froth floatation process (ii) Extraction of AI (3) Electrolytic reduction (iii) Dressing of ZnS (4) Zone refining (iv) Extraction of Au (a) (1)-(i), (2)-(ii), (3)-(iii), (4)-(iv) (b) (1)-(ii), (2)-(iv), (3)-(i), (4)-(iii) (c) (1)-(iv), (2)-(iii), (3)-(ii), (4)-(i) (1)-(iii), (2)-(i), (3)-(iv), (4)-(ii) (d) 10. Which one is used in artificial limb joints? (b) Au (a) Cu (c) Fe (d) Zn PART-II ANSWER THE FOLLOWING ANY FOUR QUESTIONS. 11. which type of ores can be concentrated by froth flotation method? give two examples 12. What are the limitations of elingham diagram? 13. What are the various steps involved in extraction of pure metals from their ores? 14. Give the uses of zinc. 15. What is auto-reduction? give example 16. What is the role of quick lime in the extraction of Iron from its oxide Fe₂O₃? PART-III ANSWER THE FOLLOWING ANY FOUR QUESTIONS. 4 X 3 = 12(COMPULSORY QUESTION NO : 22) 17. What are the differences between minerals and ores? 18. Explain the following terms with suitable example a) Gangue b) slag 19. Write a note on gravity separation method? 20. Explain concentration by magnetic separation with diagram 21. What is cementation? 22. What are the differences between Calcination and roasting? PART-IV ANSWER ALL THE QUESTION $2 \times 5 = 10$ 23.a) Explain zone refining process with an example (OR) b) Describe mond process for refining nickel. 24.a) Explain the electrometallurgy of aluminium. (OR) b) Explain froth flotation process

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