# KHADERIA HIGHER SECONDARY SCHOOL, VANIYAMBADI.

## **MODEL COMMON QUARTERLY EXAM - SEP 2024**

CLASS: 10-B SCIENCE MAXIMUM MARKS: 75

TIME: 3 HRS.

I. CHOOSE THE CORRECT ANSWER.			12 X 1=12
1. Newton's III law is applic	able		
a) For a body is at rest		b) For a body is motion	
c) both a &b		d) Only for bodies with equal masses	
2. If a substance is heated o	r cooled. The change is	mass of that substance	e is
a) Positive	b) Negative	c) Zero	d) None of the above
3. Sl unit of resistance is			
a) Mho	b) Joule	c) Ohm	d) Ohm meter
4. Which of the following ha	as the smallest mass?		
a) 6.023X 1023 atom	ns of He b) 1 a	tom of He	
c) 2g of He	d) 1 m	iole atoms of He	
5. The number of periods an	nd groups in the period	lic table are	
a) 6, 16	b) 7, 17	c) 8, 18	d) 7, 18
6. The number of componer	nts in binary solution is		
a) 2	b) 3	c) 4	d) 5
7. Rabbite belong to class			11
a) Mammal	b) Reptiles	c) Chordata	d) Amphibians
8. The wall of human hearts	s is made of		
a) Endocardium	b) Epicardium	c) Myocardium	d) All the above
9. Bipolar neurons are foun	d in		
a) Retina of eye	b) Cerebral cortex	c) Embryo	d) Respiratory epithelium
10. Which one is referred as	s 'Master gland"		
a) Pineal gland	b) Pituitary gland	c) Thyroid gland	d) Adrenal gland
11. Estrogen is secreted by		7	
a) Anterior pituitary	b) Primary follicle	c) Graffian Follicle	d) Corpus luteum
12. Okasaki fragments are j			
a) Helicase	b) DNA Polymerase	c) RNA Primer	d) DNA ligase
		T - II	
II. Answer any 7 questions. Question no. 22 compulsory.			7X2=14
13. State Newton's second l	aw.		
14. State Boyle's law.			
15. Why is tungsten metal t	ised is bulbs, but not in	fuse wires?	
16. Define Atomicity.			
17. What is rust? Give the e	-	of rust?	
18. How does leech suck blo			
19. What is the importance			
20. State whether true or fa			
	ous system is a part of	central nervous syste	m.
b) Pons helps in reg	ulating respiration.		

21. Match the following.

Column I Column II

1) Autosomes Trisomy - 21

2) Diploid condition
23rd pair of chromosome
3) Allosome
22 pair of chromosome

4) Down's syndrome - 2n

22. 3.5 litres of ethanol is present in 1.5 litres of aqueous solution of ethanol. Calculate volume percent of ethanol solution.

#### PART - III

### Answer any 7 questions. Question number 32 is compulsory.

7 X 4 = 28

- 23. List any 5 properties of light.
- 24. Distinguish between linear, areal or superficial expansion.
- 25. What is the role of earth wire in domestic circuit?
- 26. a) Define alloys. b) What are the alloys of copper and write its uses?
- 27. In what way hygroscopic substance differs from deliquescent substances.
- 28. Differentiate dicot and monocot stem.
- 29. Why is the sinoatrial node called the pacemaker of heart?
- 30. Classify neurons based on its structure?
- 31. Why did Mendel select pea plant for his experiments?
- 32. Calcium carbonate is decomposed on heating in the following reaction  $CaCO_3 \rightarrow CaO + CO_2$ 
  - i) How many moles of Calcium Carbonate are involved in this reaction?
  - ii) Calculate the gram molecular mass of Calcium Carbonate involved in this reaction.
  - iii) How many moles of CO<sub>2</sub> are there in this equation?

### PART-IV

### IV Answer all the questions. Each questions carries seven marks.

### Draw diagram wherever necessary.

7X3 = 21

33. What are the types of Inertia? Give an example for each type.

(OR

- a) What are the advantages of LED TV over the normal TV.
- b) List the merits of LED bulb.
- 34. a) Give the salient features of "Modern atomic theory'?
  - b) Give any 2 examples for hetero diatomic molecules.

(OR)

Write notes on various factors affecting solubility.

35. With a neat labelled diagram describe the parts of a typical Angiospermic ovule.

(OR)

How is the structure of DNA organised? What is the biological significance of DNA?