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Common Quarterly Examination - September 2024

Standard 10

| Time: | 3.00 | Hrs. |
|-------|------|------|
|-------|------|------|

I.

SCIENCE

Marks: 75

| Cha | aco the connect answers | | | 12×1=12 | |
|---|--|-------------------|---|----------------|--|
| | ose the correct answer: | | s of fo | rce used | |
| 1) | In which of the following sp | | | d) Hockey | |
| | a) Swimming b) Tenni | | c) Cycling | u) 11001107 | |
| 2) | Power of a lens is -4D then | | - | 0. 2.Fm | |
| | a) 4m b) –40m | | c) -0.25m | d) -2.5m | |
| 3) | The value of universal gas | constant | | | |
| | a) 3.81 J mol ⁻¹ K ⁻¹ | A. | b) 8.03 J mol ⁻¹ K | | |
| | c) 1.38 J mol ⁻¹ K ⁻¹ | | d) 8.31 J mol ⁻¹ K | -1 , # | |
| 4) | 4) Which of the following is correct? | | | | |
| a) Rate of change of charge is elecrical power. | | | | | |
| | b) Rate of change of charge | | | | |
| | c) Rate of change of energy is current. | | | | |
| | d) Rate of change of curren | it is charge. | | | |
| 5) | 5) The volume occupied by 1 mol of a diatomic gas as STP is | | | | |
| | a) 11.2 litre b) 5.6 li | tre | c) 22.4 litre | d) 44.8 litre | |
| 6) | 5) Chemical formula of rust is | | | | |
| | a) FeO×H ₂ O b) FeO ₄ | ×H ₂ O | c) Fe ₂ O ₂ ×H ₂ O | d) FeO | |
| 7) | Solubility of NaCl in 100 ml water is 36g. If 20g of salt is dissolved in 10 of water. How much more salt is required for saturation | | | | |
| | | | | | |
| | a) 12g b) 11g | | c) 16g | d) 20g | |
| 8) | Which is formed during anaerobic respiration? | | | | |
| | a) Carbohydrate b) Ethy | alcohol | c) Acetyl CoA | d) Pyruvate | |
| 9) | The body of leech has | | | | |
| | a) 23 segments b) 33 se | egments | c) 38 segments | d) 30 segments | |
| 10) Vomitting centre is located in | | | , | -, g | |
| • | a) Medulla oblongata | | b) Stomach | | |
| | c) Cerebrum | | d) Hypothalamus | | |
| 11) | | | | | |
| , | a) Jibralin b) Cyto | | c) Auxin | d) Ethylene | |
| 12) | | | • | d) Larry lene | |
| / | | | | | |
| | c) Nitrogenous base | | b) Phosphate | a to | |
| | c) Microgenous base | | d) Sugar phosph | ate | |

II. Short answer questions: [Any seven only]

7×2=14

[Question No. 15 is compulsory]

- 13) While catching a cricket ball the fielder lowers his hands backwards. Why?
- 14) What is referactive index?
- 15) How many electrons are passing per second in a circuit in which there is a current of 5A?
- 16) Give two examples of heterodiatomic molecules.
- 17) Classify the following substances into deliquescent, hygroscopic Conc, sulphuric acid, copper sulphate, penta hydrate, silicagel, calcium chloriode and gypsum salt.

- 18) What is respiratory quotient?
- 19) How are arteries and veins structurally different from one another?
- 20) What is bolting? How can it be induced artificially?
- 21) Name of the secondary sex organs in male.
- 22) What are allosomes?

III. Long answer questions: [Write any seven only] [Question No. 26 is compulsory]

 $7 \times 4 = 28$

- 23) State Newton's laws of motion.
- 24) Derive the ideal gas equation.
- 25) a) What are the advantages of LED TV over the normal TV?
 - b) Write the advantages of LED bulb.
- 26) A is a reddish brown metal, which combines with O₂ at <1370K given 13 a black coloured compound. At a temperature >1370K. A gives C which is red in colour. Find A, B and C with reaction.
- 27) In what way hygroscopic substances differ from deliquescent substances.
- 28) Describe and name three stages of cellular respiration that aerobic organisms used to obtain energy from glucose.
- 29) How does locomotion take place in leech?
- 30) What is transpiration? Give the importance of transpiration.
- 31) With a labelled diagram explain the structure of a Neuron.
- 32) a) Differentiate Monohybrid Cross and Dihybrid Cross.
 - b) What are Okazaki fragments?

IV. Answer in detail:

 $3 \times 7 = 21$

- 33) a) Describe the rocket propulsion.
 - b) List any three properties of light.

(OR)

- i) What is meant by electrical current?
- ii) Name and define its unit.
- iii) Which Instrument is used to measure the electric current? How should it be connected in a circuit?
- 34) Derive the relationship between Relative molecular mass and vapour density.

(OR)

- i) What happens when MgSO₄.7H₂O is heated? Write the appropriate equation.
- ii) Define Solubility.
- iii) Differentiate Saturated solution
 - Unsaturated solution
- 35) a) Write the physiological effects of gibberellins.
 - b) Where are the Estrogens produced? What is the role of estrogens in the human body? (OR)
 - i) With a neat labelled diagram describe the parts of a typical angiospermic ovule.
 - ii) Write significance of DNA.