18. Which is more stable Fe2+ or Fe3+? whv?

20.Calculate the PH of 0.04 M HNO3 solution .

21. what are Lewis acid and bases give one example for each ?22. How Malachitegreen is prepared from Benzaldehyde ?

19. What are interstitial compounds?

s. manikandan 770854340Me CLASS : 3.00 hrs SUBJECT: CHEMISTRY MARKS: 70 UNIT TEST - 2, 4, 8,12 `PART-I I. Choose and write the correct answer: 15X1=15 1. The most common oxidation state of actinoids is a) +2b) +3d) +6c) +42. Which one of the following ions has the same number of unpaired electrons as present in V³⁺? a) Ti³⁺ b) Fe³⁺ c) Ni²⁺ d) Cr³ 3. Which of the following fluro compounds is most likely to behave as a Lewis base? b) PF₃ c) CF₄ 4. Which one of the following statements related to lanthanons is incorrect? a) Europium shows +2 oxidation state. b) The basicity decreases as the ionic radius decreases from Pr to Lu. c) All the lanthanons are much more reactive than aluminium. d) Ce⁴⁺ solutions are widely used as oxidising agents in volumetric analysis. 5. The compound that is used in nuclear reactors as protective shields and control rods is a) Metal borides b) metal oxides c) Metal carbonates d) metal carbide 6. Which of the following is not sp² hybridised? b) graphene c) Fullerene a) Graphite d) dry ice 7. In diborane, the number of electrons that accounts for banana bonds is b) two c) four d) three 8. The pH of an aqueous solution is Zero. The solution is a) slightly acidic b) strongly acidic c) neutral d) basic 9. The pH of 10⁻⁵M KOH solution will be a) 9 c) 19 d) none of these 10. Which one of the following reduces tollens reagent a) formic acid b) acetic acid c) benzophenone d) none of these 11. The formation of cyanohydrin from acetone is an example of a) nucleophilic substitution b) electrophilic substitution c) electrophilic addition d) Nucleophilic addition 12. Which one of the following reaction is an example of disproportionation reaction a) Aldol condensation b) cannizaro reaction c) Benzoin condensation d) none of these 13.ln ____ carbon atoms are arranged in the shape of a football a) fullerenes b) graphene c) carbon nanotube d) diamond show the highest oxidation state of +8 b) Os and Mn a) Ru and Mn d) Ru and Os c) Mn and Cr 15. The order of reactivity of carboxylic acid derivatives is _ a) Acid halide > Ester > Amide > Acid anhydrides b) Acid halide > Acid anhydrides > Ester > Amide c) Acid halide > Amide > Acid anhydrides > Ester d) Acid anhydrides > Ester > Amide > Acid halide PART-II II. Answer any six questions (Q.no.24 is compulsory) $6 \times 2 = 12$ 16. What is inert pair effect? 17. How will you convert boric acid to boron nitride?

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- 23. Give one example for the followin UNIT TEST 2, 4, 8,12 i) icosagen ii) chalcogen iii) tetragen iv) Pnictogen
- 24. Write the expression for the solubility product of Hg₂Cl₂ and BaSO₄

PART-III

III. Answer any six questions (Q.no.33 is compulsory)

 $6 \times 3 = 18$

- 25. Write ethyl borate test?
- 26. What is catenation? Write any two conditions for catenation?
- 27.Write chromyl chloride test?
- 28. Write a note on Fischer tropsch synthesis?
- 29. Explain why Cr²⁺ is strongly reducing while Mn³⁺ is strongly oxidizing.
- 30. Explain common ion effect with an example
- 31. Write the test for carboxylic acid group?
- 32. Explain the reducing action of formic acid with example
- 33. Identify A and B



IV. Answer all the questions .

- 34.a) i) AlCl₃ is more stable. TlCl₃ is less stable why ?(2)
 - ii) Write the difference between graphite and diamond ?(3)

- b) Describe the structure of Diborane
- 35. a) What is lanthanoid contraction and what are the consequences of lanthanoid contraction? (OR)
 - b) i) What are the properties of interstitial compounds? (3)
 - ii) why d block elements exhibit variable oxidation state?(2)
- 36. a) Derive an expression for ostwald dilution law

(OR)

- b) i) Derive henderson equation(3)
 - ii) What are the limitations of Arrhenius concept? (2)
- 37. a) i) What is formalin? what is its use(2)
 - ii) how will you convert benzaldehyde into the following compounds?(3)
 - i) benzoin

ii)cinnamic acid

- b) Write the mechanism of aldol condensation reaction
- 38. a) Compare properties of lanthanides and actinide

(OR)

- b) i) Define buffer Index(2)
 - ii) Write a note on Rosenmund reduction ?(3)

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