ONE MARK REVISION TEST

Class: XI Code: A1118 Max. Marks: 25

11. FUNDAMENTALS IN ORGANIC CHEMISTRY

I. Choose the correct Answer

 $25 \times 1 = 25$

- 1. How many cyclic and acyclic isomers are possible for the molecular formula C3H6O?
 - a) 4
- b) 5
- c) 9
- d) 10
- 2. Which one of the following shows functional isomerism?
 - a) ethylene
- b) Propane
- c) ethanol
- d) CH₂Cl₂
- 3. Which of the following names does not fit a real name?
 - a) 3 Methyl –3–hexanone c) 3– Methyl –3– hexanol
 - b) 4–Methyl –3– hexanone
- d) 2- Methyl cyclo hexanone.
- 4. The IUPAC name of CH₃-CH=CH-C≡CH is
 - a) Pent-4-yn-2-ene
- c) pent-2-en-4- yne
- b) Pent-3-en-l-yne
- d) Pent -1-yn-3-ene
- 5. Hybridization of underlined carbon in allene compound is

$$CH_2 = \underline{C} = CH_2$$

- a) sp
- b) sp2
- c) sp3
- d) dsp2
- 6. The general formula for alkadiene is
 - a) C_nH_{2n}
- b) C_nH_{2n-1}
- c) C_nH_{2n-2}
- d) C_nH_{n-2}
- 7. The IUPAC name of the Compound is

$$CH_3$$
 CH_3
 CH_3

- a) 2,3-dimethylheptane
- c) 5-ethyl-6-methyloctane
- b) 3-Methyl-4- ethyloctane
- d) 4-Ethyl -3 methyloctane
- 8. In an organic compound, phosphorus is estimated as
 - a) Mg₂P₂O₇
 - b) Mg₃(PO₄)₂ c) H₃PO₄
- d) P₂O₅

9. Consider the following statement.

- a) Pentanal does not show position isomerism.
- b) Meso tartaric acid is optically active in nature.
- c) Ethanol is functional isomers with Dimethyl ethers.

Which of the above statements is/are correct?

a) (ii) only

c) Both (i) and (iii)

b) Both (i) and (ii)

d) none of the above

10. Nitrogen detection in an organic compound is carried out by Lassaigne's test. The blue colour formed is due to the formation of.

a) Fe₃ [Fe(CN)₆]₂

c) Fe₄ [Fe (CN)₆]₂

b) Fe₄ [Fe (CN)₆]₃

d) Fe₃ [Fe (CN)₆]₃

11. Which of the following is optically active alcohol?

a) 1-propanol

c) 1-butanol

b) 2-propanol

d) 2-butanol

12. Lassaigne's test for the detection of nitrogen fails in

- a) H₂N-CO-NHNH₂.HCl
- c) C₆H₅-NHNH₂.HCl

b) H₂NNH₂.HCl

d) C₆H₅ CONH₂

13. Match the following:

Functional Group			Compound				
A	-СНО	1	Propanone				
В	-NO ₂	2	Acetamide				
С	>C=O	3	Ethanal				
D	-CONH ₂	4	Nitrobenzene				

	A	B	C	D		A	В	C	D
a)	3	4	1	2	c)	3	1WW	4	2
b) (e)	4	1	2	3	d)	4 10	3	1	2

14. Connect pair of compounds which give blue colouration / precipitate and white precipitate respectively, when their Lassaigne's testis separately done.

- a) NH2NH2 HCl and ClCH2-CHO
- b) NH₂CSNH₂ and CH₃-CH₂Cl
- c) NH2CH2COOH and NH2CONH2
- d) $C_6H_5NH_2$ and $ClCH_2$ -CHO.

15. Select the molecule which has only one π -bond.

- a) CH₃CH = CHCH₃
- c) CH₃CH = CHCOOH
- b) CH₃CH = CHCHO
- d) All of these

16. Number of stereoisomers of 1, 2-dihydroxycyclopentane

- a) 1
- b)2
- c) 3
- d) 4

17. Sodium nitroprusside reacts with Sulphide ion to give a purple colour due to the formation of

a) [Fe(CN)₅ NO]³-

c) [Fe(CN)₅NOS]⁴-

b) [Fe(NO)₅ CN]⁺

d) [Fe (CN)5 NOS]3-

18. Ortho and para-nitro phenol can be separated by

- a) Azeotropic distillation
- c) Steam distillation
- b) Destructive distillation
- d) Cannot be separated

19. The purity of an organic compound is determined by

- a) Chromatography
- c) melting or boiling point

b) Crystallisation

d) both (a) and (c)

20. <u>Assertion:</u> Organic compound is given below named as 3-carbethoxy-2-butenoicacid

$$CH_3 - C = CH - COOH$$
 $COOC_2H_5$

Reason: Principal functional group gets lowest number followed by double bond (or) triple bond.

- a) Both assertion and reason are true and reason is the correct explanation of assertion.
- b) Both assertion and reason are true but reason is not the correct explanation of assertion.
- c) Assertion is true but reason is false.
- d) Both assertion and reason are false.

21. Fill in the blank:

pentan-2-one; pentan-2-one; 3-methylbutan-2-one 1-methoxypropane; 2-methoxypropane;

a) Dimethyl ether

c) Dimethyl ketone

b) Diethyl ether

d) 3-methoxypropane

22. The compound that shows geometrical isomer is ____

- a) 1-propene b) 1-butene
- c) 2-butene
- d) Both (a) and (b)
- 23. A sample of 0.5g of an organic compound was treated according to Kjeldahl's method. The ammonia evolved was absorbed in 50 mL of 0.5M H₂SO₄. The remaining acid after neutralization by ammonia consumed 80 mL of 0.5 M NaOH, % of nitrogen in the organic compound is.
 - a) 14%
- b) 28%
- c) 42%
- d) 56%
- 24. The oxidation state of C in CH₂Cl₂ is
 - a) 0
- b) 1 (et
- c) 2
- d) 4
- 25. From the following organic compounds, find the Aromatic non-benzenoid compound:







