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MODEL QUESTION PAPER **CLASS: X STD SCIENCE MAX: 100 MARKS** UNIT: 1, 2, 3,7,8,9,12,13,14 PHYSICAL SCIENCE I. CHOOSE THE BEST ANSWER 1. Power of a lens is 4D, then its focal length is a) 4m b) -40m c) -0.25 m d) 0.25 m 2) Impulse is equals to a) rate of change of momentum b) rate of force and time c) change of momentum d) rate of change of mass 3. If a substance is heated or cooled, the change in mass of that substance is a) positive b) negative c) zero d) none of the above 4. In the nucleus of 20Ca<sup>40</sup>, there are a. 20 protons and 40 neutrons b. 20 protons and 20 neutrons c. 20 protons and 40 electrons d. 40 protons and 20 electrons 5. Neon shows zero electron affinity due to a) stable arrangement of neutrons b) stable configuration of electrons c) reduced size d) increased density 6. A 25% alcohol solution means a. 25 ml alcohol in 100 ml of water b. 25 ml alcohol in 25 ml of water c. 25 ml alcohol in 75 ml of water d. 75 ml alcohol in 25 ml of water II ANSWER THE FOLLOWING QUESTIONS Q.NO: 10 IS COMPULSORY  $5 \times 2 = 10$ 7. While catching a cricket ball the fielder lowers his hands backwards. Why? 8. State Snell's law. 9. State-the law of volume 10. Two blocks of masses 8 kg and 2 kg respectively lie on a smooth horizontal surface in contact with one other. They are pushed by a horizontally applied force of 15 N. Calculate the force exerted on the 2 kg mass. (or) ii) 1.51x 10<sup>23</sup> molecules of NH4Cl Calculate the number of moles in i) 27g of Al 11. a) Define: Atomicity b) State two conditions necessary for rusting of iron. 12. a) Match the following 1. Blue vitriol - CaSO4 .2H2O 2. Gypsum - CaO - CuSO4.5H2O 3. Deliquescence 4. Hygroscopic - NaOH B) What is mean by binary solution. III ANSWER THE FOLLOWING QUESTIONS Q.NO: 17 IS COMPULSORY  $5 \times 4 = 20$ 13. What are the types of inertia? Give an example for each type. 14. Explain the experiment of measuring the real and apparent expansion of a liquid with a neat diagram. 15. a) The aquatic animals live more in cold region Why? b) Define modern periodic law. 16. Relationship between vapour density and relative molecular mass 17. How many grams are there in the following? i. 2 moles of hydrogen molecule, H2 ii. 3 moles of chlorine molecule, Cl2 iii. 5 moles of sulphur molecule, S8 iv. 4 moles of phosphorous molecule, P4 (or) An object of height 3cm is placed at 10cm from a concave lens of focal length 15cm. Find the size of the image. IV ANSWER IN BRIEF 2 X 7 = 1418. (a) Define coralent radius (b) What is innertransition elements (c) What is periodicity 19. i) State and prove the law of conservation of linear momentum.

ii) How does an astronaut float in a space shuttle?

a) Differentiate the eye defects: Myopia and	Hypermetropia	MNNA	MANIA
b) Why are traffic signals red in colour?			
20. A) Give the salient features of "Modern ator	nic theory". (3m)		
B) write the difference between minerals	and ores.(2m)		
c) Define Hydrated salt. (2m)			
	(or)		
21 i) Write notes on i) saturated solution ii) u	nsaturated solution (3m)		
ii) What is rust? Give the equation for form	ation of rust. (2m)		
iii) Give any two examples for hetero diator	nic molecules. (2m)		
	BIOLOGY		
	SECTION - A		
I.Choose the correct answer		6x1=6	
1. The endarch condition is the characteristic fea	ature ofa)root b)ste	m c)leavesd)flower	
2. Which is formed during anaerobic respiration	My Hall .	My House	
a)carbohydrate b)ethyl alcoholc)acety			
3.The brain of leech lies above the a)Mout		Pharynx d)Crop	
4. The animals which give birth to young one	22939510	padasetro	
	vi parous d)a	all the above	
5.During transpiration there is loss of	IV/2 v .		
6.Which of the following process requires energ		970:	
A)active transport b)diffusion	c)osmosis d)all of the	m usalala	
2009 Thomas of the party of the	. Page a a a a a a a a a a a a a a a a a a	Pages	
	SECTION - B		
IIAnswer briefly: Q.No : 10 is compulsory	DECTION B	5x2=10	
7.What is respiratory quotient?		3,22-10	
8. Write the reaction for photosynthesis.			
9. Why are the rings of cartilages found in trache	a of rabbit?		
10. How are arteries and veins structurally differ			
11. What is the importance of valves in the heart			
12. Why is the circulation in man reffered to as of			
13.What is cohesion?	iouble circulation:		
13.What is collesion?			
III Angway in payagraph O No . 10 is compuls	OW.	5x4=20	
III.Answer in paragraph.Q.No: 18 is compuls		3x4=20	
14.Differentiate between systemic circulation	10000		
15. What is photosynthesis? where in a cell does	s it occur?		
16.Differentiate monocot and dicotroot			
17.List out the parasitic adaptative in leeeh			
18.Explain the functions of mitochondria.			
19. How does locomotion take place in leech?			
20.Write the importance of transpiration?	WWW ST.		
	SECTION - D	4-70	
IV.Answer in Detail	0190	2x7=14	9
21. Describe and name three stage s of calculate	respiration that aerobic org	ganism use to obtain energy	trom
glucose.	WAR PARKET		
	OR		
Explain the male reproductive system of rab	7 1961		
22. Why are leucocytes are classified as granulo	= ~2/19/10/1		
Name each cell and mention it's function	(OR)	Enumerate the functions of	f blood.



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